



## **Temperance Term**

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7	Week 8	
Topic	C9 Chemistry of the Atmosphere				Revision: C1- Atomic Structure and the Periodic Table	RM			
Challenge Objective and Content (for all learners)	C9-Describe and explain the factors which can affect the chemistry of the Earth's atmosphere  -Describe the Earth's early atmosphere  -Describe the main changes in the atmosphere over time and the likely cause of these changes  -Describe the greenhouse effect			C10-Explain ways to u -Distinguish between f -Describe the difference -Carry out a life cycle a	•	C1 – Describe the development and structure of the atom and periodic table.	HALF TE		
Inspire Opportunities	Evaluate the quality of evidence in a report about global climate change			Evaluate alternative methods of metal extraction				Grade 7-9 questions	
Assessment Opportunities	End of Topic Tests			End of Topic Tests				Exam Questions	

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	
Торіс	Revision: C2- Bonding, Structure and the Properties of Matter	Revision: C3- Quantitative Chemistry	Revision: C4- Chemical Changes	Revision: C5- Energy Changes	Mocks (F	Paper 1)	IAS
Challenge Objective and Content (for all learners)	C2 – Explain how atoms are held together in structures and describe their properties.	C3 – Use chemical equations as a way to communicate chemical ideas.	C4 – Investigate and predict chemical changes in substances	C5-Explain how the interaction of particles often involves transfers of energy.			CHRISTMA
Inspire Opportunities	Grade 7-9 questions	Grade 7-9 questions	Grade 7-9 questions	Grade 7-9 questions			
Assessment Opportunities	Exam Questions	Exam Questions	Exam Questions	Exam Questions			





### **Justice Term**

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	
	Revision: C6-	Revision: C7-	Revision: C8 – Ch	emical Analysis	Revision: C9	Revision: C10 -	
	The Rate and	Organic			Chemistry of the	Using	
Topic	Extent of	Chemistry			Atmosphere	Resources	
	Chemical						
	Change					T	5
Challenge Objective and Content (for all learners)	C6-Understand energy changes that accompany chemical reactions.	C7-Explain the importance of carbon compounds as organic compounds, in terms of structure and properties.	C8-Explain a variety of instrumental methods can be used to analyse substances		C9-Describe and explain the factors which can affect the chemistry of the Earth's atmosphere	C10-Explain ways to use the Earth's natural resources to manufacture useful products	HALF TERM
Inspire Opportunities	Grade 7-9 questions	Grade 7-9 questions	Grade 7-9	questions	Grade 7-9 questions	Grade 7-9 questions	
Assessment Opportunities	Exam Questions	Exam Questions	Exam Qı	uestions	Exam Questions	Exam Questions	

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	
Topic	Mocks (Paper 2)	Revision: C1- Atomic Structure and the Periodic Table	Revision: C2- Bond the Propertie	ling, Structure and es of Matter	Revision: C3- Quantitative Chemistry		STER
Challenge Objective and Content (for all learners)		C1 – Describe the development and structure of the atom and periodic table.	C2 – Explain how atom structures and describ	•	C3 – Use chemical equ communicate chemica	•	EA





Inspire Opportunities	Grade 7-9 questions	Grade 7-9 questions	Grade 7-9 questions	
Assessment Opportunities	Exam Questions	Exam Questions	Exam Questions	





## **Courage Term**

	Week 1	Week 2	Week 3	Week 4	Week 5		
Торіс	Revision: C4- Chemical Changes		Revision: C5- Er	ergy Changes	Science Skills	_	
Challenge Objective and Content (for all learners)	C4 – Investigate and predict chemical changes in substances		C5-Explain how the i particles often involvenergy.			HALF TERM	
Inspire Opportunities	Grade 7-9 questions		Grade 7-9 questions		Grade 7-9 questions	H	
Assessment Opportunities	Exam Questions		Exam Qu	estions	Exam Questions Chemistry Paper 1 - 19 <sup>th</sup> May		

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	
Торіс	,		Study	Leave			
Challenge Objective and Content (for all learners)							SUMMER
Inspire Opportunities							S
Assessment Opportunities	Chemistry Paper 2 – 1	13 <sup>th</sup> June					